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Two unusual minor 18,19-seco-ursane glycosides from leaves of Ilex cornuta

Tao Wu^{a,*}, Yan Li^a, Qing-jiu Tang^b, Zheng-tao Wang^a

^a Key Laboratory of Standardization of Chinese Medicines, Ministry of Education, Institute of Chinese Materia Medica, Shanghai University of Traditional Chinese Medicine, 1200 Cailun Road, Shanghai 201203, China

^b Shanghai Edible Fungi Institute, Shanghai Academy of Agricultural Sciences, Shanghai 201106, China

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ABSTRACT

The leaves of *llex cornuta* is traditionally used as a functional tea in China. Two new minor 18,19-*seco*ursane glycosides, named cornutaoside A (1) and B (2), were isolated from leaves of *l. cornuta*, along with two known compounds (3 and 4). Their structures were elucidated as $(3\beta,12\beta)$ -3- $[\beta$ -D-glucopyranosyl- $(1 \rightarrow 2)$ - α -L-arabinopyranosyl]-12,21-dihydroxy-19-oxo-18,19-secours-13(18)-en-28-oic acid (1) and $(3\beta,12\beta)$ -3- $[\beta$ -D-glucopyranosyl- $(1 \rightarrow 2)$ - α -L-arabinopyranosyl]-12,19,21-trihydroxy-18,19-secours-13(18)-en-28-oic acid (2), by chemical methods, 1D and 2D NMR experiments, and by comparison with known analogues. This is the first report of *E-seco* triterpenoids and diterpene skeletons (4) from this plant. In a preliminary cytotoxic test against U937, L1210, and B16 cell lines, 1 and 2 had no significant activities as compared to controls, with concentrations up to 443.61 and 346.25 μ M/plate, for 1 and 2, respectively.

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1. Introduction

Leaves of some *llex* species are used to make a widely consumed beverage in China called "Kudingcha", bitter tea (Chau & Wu, 2006; Heck & Mejia, 2007; Jiangsu New Medical College, 1986). In addition to its common consumption, Kudingcha is also popular for its health benefits. Ilex cornuta Lindl. et Paxt. (Aquifoliaceae) is an evergreen shrub growing in Eastern and Southern China. The dried leaves of *I. cornuta* are traditionally employed as one of the staple Ilex species for "Kudingcha" or as an ingredient in dietary supplements. In addition, the aqueous decoction made from its leaves is commonly used as a contraceptive, cardiovascular or antibacterial agent (Jiangsu New Medical College, 1986; Qin et al., 1986; Zhang, Lin, Cen, & Shen, 2003). Previous phytochemical investigation of I. cornuta showed that leaves of this plant are a rich source of triterpenoids and flavonoids as well as their corresponding glycosides (Li, Wu, Cheng, & Wang, 2006; Nakanishi, Terai, Nasu, Miura, & Yoneda, 1982; Oin et al., 1986; Wu, Cheng, Liu, Li, & Wang, 2005). The saponin-containing fractions from leaves of this plant have been reported to possess various important pharmacological effects such as the increase of coronary blood flow and anti-hematoblastic coagulation activities (Qin, Zhao, & Fukuyama, 1988; Qin et al., 1986). Recently, our investigation of leaves of this plant demonstrated again that triterpenoids were the predominant components among its secondary metabolites (Li et al., 2006; Wu et al., 2005).

As part of our continuous search for potentially active substances from *I. cornuta*, we report herein the isolation and structural elucidation of two unusual minor 18,19-*seco*-ursane glycosides along with a known triterpenoid saponin and a known diterpene. This is the first report of 18,19-*seco*-triterpenoids and diterpene skeletons from this species.

2. Materials and methods

2.1. General

Optical rotations were measured with a JASCO P-1030 polarimeter (Jasco Co., Tokyo, Japan). The IR spectra were obtained on a Perkin-Elmer Spectrum One FR-IR instrument (Perkin-Elmer Co., Norwalk, CT, USA). NMR spectra were recorded on Bruker ACF-500 spectrometers (Bruker Instruments, Inc., Billerica, MA, USA) $[^{1}H (500 \text{ MHz}) \text{ and } ^{13}C (125 \text{ MHz})]$ in pyridine- d_{5} as a solvent. Chemical shifts were reported using residual pyridine ($\delta_{\rm H}$ 7.18 and $\delta_{\rm C}$ 123.5) as internal standard. ESIMS and HR-Fourier Transformation ESIMS were obtained on a Bruker APIII MS instrument (Bruker Daltonics, Billerica, MA, USA) equipped with Microvip data system. HP-TLC was performed on silica gel 60 F₂₅₄ plates (Merck, Darmstadt, Germany). Reversed-phase preparative HPLC separation was carried out on an Agilent 1100 apparatus equipped with a 150 \times 4.6 mm i.d. C_{18} silica gel column (ZORBAX SB-Aq, Agilent Co. Ltd., Palo Alto, CA, USA) at 25 °C and eluted with CH₃CN-H₂O (37:63, v/v, 1.0 mL/min) as mobile phase, using a G1315A DAD detector (Agilent Co. Ltd., Palo Alto, CA, USA).





^{*} Corresponding author. Tel.: +86 21 51322513; fax: +86 21 51322519. *E-mail address*: laurawu2000@hotmail.com (T. Wu).

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2.2. Plant material

The leaves of *I. cornuta* were collected from Shanghai, China, in August 2001, and identified by Prof. Zheng-tao Wang. A voucher specimen (Herbarium No. 0115) is on deposit at the herbarium of the present institute.

2.3. Extraction and isolation

The shade-dried and powdered leaves (10 kg) of I. cornuta were extracted with 85% aqueous ethanol (50 L \times 2) at room temperature, and partitioned between *n*-butanol and H₂O. The *n*-butanol extract (117 g) was chromatographed over silica gel column by eluting with a gradient of CH₂Cl₂/CH₃OH (20:1-1:1, v/v) to afford five fractions I-V as described previously (Li et al., 2006). Compound 4 (88 mg) was obtained by separation of fraction I on silica gel with $CH_2Cl_2/CH_3OH(15:1, v/v)$ as elutant. Separation of fraction V on silica gel column eluted by increasing amounts of CH₃OH in EtOAc afforded three subfractions: subF₁, F₂ and F₃. A portion of subF₂ was further purified by repeated silica gel column (CH₂Cl₂/ MeOH, 95:4, v/v) followed by preparative HPLC on RP-18 (CH₃CN/H₂O, 37:63, v/v) to afford compounds 1 (40 mg) and 2 (35 mg). The subF₃ was subjected to repeated silica gel column eluting with CH₂Cl₂/MeOH (5:1, v/v) to afford compound 3 (100 mg).

2.4. Acidic hydrolysis and GC analysis of compounds 1 and 2

Each compound (10 mg) was heated in 2 M trifluoroacetic acid at 120 °C for 2 h. The reaction mixture was partitioned with EtOAc. The aqueous layer was concentrated, and the residue was re-dissolved in MeOH and subjected to HP-TLC analysis with authentic sugars, applying silica gel HP-TLC plates, using EtOAc/H₂O/formaic acid/acetic acid (100:27:11:11, v/v/v/v) as the developing reagent. Spot of TLC was visualized by spraying with 10% H₂SO₄ in EtOH followed by heating at 110 °C for 5 min. Acidic hydrolysis of both 1 and **2** gave glucose and arabinose. The absolute configuration of glucose and arabinose was determined according to the method described by Cases, Cerezo, and Stortz (1995). Briefly, the hydrolyzates of 1 (7.0 mg), 2 (6.7 mg), or sugar standards (10.0 mg) were weighed into 25 mL glass-stoppered flasks. The following anhydrous methanolic solutions were sequentially added to the samples: (i) 11.11% (S)-1-amino-2-propanol (20 µL of solution/mg sugar); (ii) 20% glacial AcOH (17 µL of solution/mg sugar); and (iii) 3% NaBH₃CN (13 µL of solution/mg sugar). The resulting mixture was heated and reacted for 1.5 h at 65 °C. Then the reaction mixture was cooled to room temperature, and 3 M aqueous CF₃COOH was added drop wise until the pH was 1–2. The mixture was evaporated, followed by co-evaporation with addition of water $(3 \times 0.5 \text{ mL})$ and MeOH $(5 \times 0.5 \text{ mL})$. Trace solvent was further removed overnight in a desiccator. Two mL of 50% pyridine in Ac₂O was added to the residue, and the mixture was heated for 45 min at 100 °C, followed by partition with EtOAc and washing with water (3 \times 1.0 mL) and saturated NaHCO₃ (3 \times 1.0 mL). The organic phase was dried with anhydrous Na₂SO₄ and subjected to GC analysis [TRACE GC ULTRA-DSQ (Thermo, San Jose, CA) equipped with a mass detector and a TRB-5MS capillary column $(30 \text{ m} \times 0.25 \text{ mm} \times 0.25 \text{ }\mu\text{m})]$. The oven temperature program was initiated at 130 °C (held for 2 min), then raised at 2 °C/min to 159 °C (held for 7 min), followed by a rise at 2 °C/min/ to 169 °C for 3 min, 175 °C for 2 min, 179 °C for 2 min, 185 °C for 2 min, 187 °C for 2 min, and finally, at 10 °C/min to 205 °C (held for 10 min). The identification of the sugars was achieved by comparison of their retention times (R_t : 59.84 and 60.25 min for L-arabinose and D-glucose, respectively) with those of authentic sugars. D-Glucose and L-arabinose were detected for compounds **1** and **2**.

2.5. Cytotoxic assay

The U937, L1210 and B16 cell lines were cultured, respectively, in Dulbecco's modified Eagle medium (180 μ L; *Gibco BRL*) in 96well micro-culture plates according to a previously described method (Wu et al., 2005). Medium (20 μ L) containing different concentrations of compound **1** or **2** were added to all wells (in triplicate). Control cells received medium containing analogous dimethyl sulfoxide (DMSO) concentrations. After incubation at 37 °C in a 5% CO₂ atmosphere for three days, Alamar Blue (20 μ L) was added to each well. After incubation for another 6 h, the color change of the solution was measured with an ELISA micro-autoreader at 570 and 600 nm. The activation rate was calculated according to the formula provided by the Alamar Blue assay (*Nivelles*, Belgium).

3. Results and discussion

3.1. Structure elucidation of the novel 18,19-ursane glycosides

Solvent extraction, followed by repeated silica gel column chromatography, and preparative high performance of liquid chromatography, of aqueous ethanol extract of *I. cornuta*, led to isolation of two 18,19-*seco*-ursane glycosides (Fig. 1) and two known compounds.

Compound **1** was obtained as colorless solids from MeOH, $[\alpha]_D^{20}$ - 29.1 (c 0.39, MeOH), and its molecular formula was determined as $C_{41}H_{66}O_{15}$ on basis of the quasi-molecular ion observed at m/z803.4184 (HR-ESIMS, $[(M-H_2O) + Na]^+$, calcd. for $C_{41}H_{64}NaO_{14}$, 803.4188), and confirmed by ¹³C NMR (DEPT) analysis. The IR spectrum exhibited strong absorptions at 3414, 1751, 1719 and 1624 cm⁻¹, indicating the existence of hydroxy, $2 \times$ carbonyl, and olefinic functionalities. The ¹H NMR spectrum of **1** showed characteristic signals of triterpenoids with seven tertiary methyl signals at δ 0.89 (3H, s), 1.06 (3H, d, I = 7.1 Hz), 1.09 (3H, s), 1.31 (6H, s, overlap), 1.37 (3H, s) and 2.24 (3H, s), a typical 3α -H signal at δ 3.34 (1H, dd, J = 11.7, 4.3 Hz), two oxygen-bearing methine protons at δ 4.69 (1H, t, J = 6.8 Hz) and 4.92 (1H, m), and an olefinic proton at δ 6.24 (1H, s), as well as two anomeric signals at δ 5.06 (br s, overlap) and 5.27 (1H, d, J = 7.7 Hz). The ¹³C NMR spectrum of **1** showed 41 carbon signals, with 30 attributed to the aglycone part and 11 to two monosaccharides (Table 1). Note that two diagnostic olefinic carbon signals observed at δ 151.16 and 119.64, the characteristic resonances of H-12 [δ_C 69.34, δ_H 4.69 (1H, t, J = 6.8 Hz)], and the C-30 methyl [δ_{C} 11.91, δ_{H} 1.06 (1H, d, J = 7.1 Hz)] geminal to a hydroxyl, were characteristics of 18,19-seco-ursane derivatives, which possess C-13(18) double bond and β -hydroxyls at C-12 or C-21 (Kakuno, Yoshikawa, Arihara, Takei, & Endo, 1991; Li et al., 2005). Moreover, existence of two hydroxyls attributed to HO-12 and HO-21 were supported by the fragments at m/z 803 $[(M-H_2O) + Na]^+$ and 785 $[(M-2H_2O) + Na]^+$ present in the positive ESI mass spectrum. Complete assignments of 1 based on a detailed analysis of ¹H, ¹³C NMR, DEPT, HMQC, and HMBC allowed establishing the exact nature of the aglycone of **1**, which is in good agreement with that reported for 18.19-seco-ursane (Li et al., 2005). Further structural confirmation was performed by combinational analysis of HMQC and HMBC spectra. As shown in Fig. 2, the three-bond correlations of H-9 and H-18 with C-12 confirmed the position of HO-12, while the long-range correlations of H-22, H-20 and Me-30 with C-21 indicated the presence of HO-21. In addition, the determination of endocyclic double bond at C-13(18) and the C-19 ketone group were verified by HMBC correlations of H-18



Fig. 1. Structures of compounds 1 and 2.

Table 1 ¹³C NMR data of compounds **1** and **2** (125 MHz, pyridine- d_5)

No.	Aglycone moiety		No.	Sugar moiety	
	1	2		1	2
1	39.25	39.26	Inner arab-		
2	26.72	26.74	1′	104.98	104.98
3	88.89	88.91	2′	81.21	81.21
4	39.75	39.75	3′	73.54	73.55
5	56.17	56.19	4′	68.42	68.41
6	18.49	18.52	5′	65.08	65.08
7	35.20	33.16	Terminal glc-		
8	41.42	41.45	1″	106.15	106.16
9	49.92	49.98	2''	76.53	76.54
10	37.43	37.44	3′′	78.29	78.31
11	33.10	35.24	4''	71.78	71.78
12	69.34	69.37	5''	78.35	78.36
13	151.16	150.66	6′′	62.76	62.76
14	43.47	43.50			
15	27.72	27.82			
16	29.54	29.78			
17	45.29	45.78			
18	119.64	120.20			
19	209.45	68.38			
20	51.94	45.61			
21	77.62	77.97			
22	43.23	43.93			
23	28.34	28.35			
24	16.80	16.80			
25	16.88	16.90			
26	18.94	21.60			
27	22.33	22.39			
28	178.82	179.52			
29	29.65	19.01			
30	11.91	10.43			

with C-14, C-16, and C-17, and of Me-29, Me-30, and H-20 with C-19, respectively.

The nature of the sugars (glucose and arabinose) of **1** was first determined by the comparison of their ¹³C NMR data (Table 1) with the corresponding monosaccharides (Yu & Yang, 1999), which was further confirmed by the high performance-TLC (HP-TLC) analysis with authentic samples after acidic hydrolysis. To verify their absolute configurations, **1** was subjected to acidic hydrolysis, followed by GC analysis in comparison with D-glucose and L-arabinose (Cases et al., 1995). By this procedure the sugars were identified as D-glucose and L-arabinose. The sequence of the monosaccharide residues was deduced from the presence of cross-peaks in HMBC

spectra (Fig. 2). The glycosidic linkage at C-3 of the aglycone was shown by the HMBC correlation of the anomeric proton at δ 5.06 (br s, overlap) with C-3 at δ 88.89. The 1 \rightarrow 2 interglycosidic linkage with a terminal D-glucose was assigned by the HMBC correlation between H-1" of D-glucose and C-2' of L-arabinose as well as the comparison of ¹³C chemical shifts for each methylated monosaccharide (Yu & Yang, 1999). In addition, this assignment of the β -D-glucopyranosyl (1 \rightarrow 2)- α -L-arabinopyranosyl unit was further supported by the reported ¹³C NMR data of a similar disaccharide chain (Nakanishi et al., 1982).

The relative configuration of **1** was proposed as shown in Fig. 1 by analysis of NOESY data and comparison of the documented data. As mentioned above, the superimposable NMR signals of the aglycone of **1** and the reported 18.19-seco-ursane derivatives (Kakuno, Yoshikawa, & Arihara, 1991: Li et al., 2005), indicated the B-orientation of Me-24, Me-25, Me-26, HO-12, and COOH-28, and α-orientation of H-3, H-5, H-9 and Me-27. However, limited cross-peak information was obtained in several trials of NOESY experiments, except that correlations between H-3 and H-5, H-3 and H-1', H-5 and H-6 α , and Me-25 and Me-26 confirmed that the configurations of H-3 and H-5 were α , and that of Me-25 and Me-26 were β . Note that the relative configurations at C-20 and C-21 were not ascertained yet due to the free rotation of the σ -bond between C-17 and C-22, as suggested by Li et al. (2005). Based on the above evidence, the structure of **1** was established as $(3\beta, 12\beta)$ -3- $[\beta$ -Dglucopyranosyl- $(1 \rightarrow 2)$ - α -L-arabinopyranosyl]-12,21-dihydroxy-19-oxo-18,19-secours-13(18)-en-28-oic acid, named cornutaoside A. The detailed NMR spectra data of cornutaoside A are listed in Tables 1 and 2.

Compound **2** was obtained as a white amorphous powder, $[\alpha]_D^{20}$ -26.3 (*c* 0.34, MeOH). It exhibited a pseudo-molecular ion peak at *m*/*z* 805.4392 ([(M–H₂O) + Na]⁺, calcd. for C₄₁H₆₆O₁₄Na, 805.4404) in HR-ESIMS, consistent with a molecular formula of C₄₁H₆₈O₁₅ assigned by a combinational analysis of ¹H, ¹³C NMR and DEPT. Inspection of ¹H, ¹³C NMR data of **2** revealed that compounds **1** and **2** were closely comparable (Tables 1 and 2). The superimposable signals included seven tertiary methyl signals at δ 0.90 (3H, s, Me-25), 1.09 (6H, Me-24 and 30), 1.28 (6H, Me-23 and 27), 1.36 (3H, s, Me-26) and 1.41 (3H, d, *J* = 6.1 Hz, Me-29), a 3 α -H signal at δ 3.33 (1H, dd, *J* = 11.0, 3.1 Hz), and H-18 olefinic proton at δ 6.29 (1H, s), two oxygen-bearing methine protons of HO-12 and HO-21 at δ 4.69 (1H, t, *J* = 6.2 Hz) and 5.16 (1H, m), as well as two anomeric signals at δ 5.05 (br s, overlap) and 5.27



Fig. 2. Key HMBC correlations for cornutaoside A (1) and cornutaoside B (2).

(1H, d, J = 7.7 Hz). The significant difference which appeared in ¹H NMR was that the ketone group in **1** was replaced with a hydroxyl group in **2** at C-19 position (δ 4.10). The comparison of ¹³C NMR data of 1 and 2 showed that differences appeared only in the resonances associated with the 18,19-seco-ursane section at C-19 to C-22, which replacement of the ketone group at C-19 in **1** by an additional hydroxyl at δ 68.38 in **2** indicated a 19,21-diol structure of that chain. This assignment was established by ¹³C NMR signals of C-19, C-20, C-21, and C-22, observed at δ 68.38, 45.61, 77.97, and 43.93, respectively. Additional evidence for this assignment was obtained by the HMBC correlations of C-19 with Me-29 and Me-30 (Fig. 2), and an ESI quasi-molecular ion peak at m/z 805 $[(M-H_2O) + Na]^+$, differing from 1 by 2 amu. The sugar part of 2 was determined as the same with that of **1** by the comparison of the NMR data (Table 1) and the methods applied to 1 including HMBC experiments, chemical hydrolysis, and GC analysis (Fig. 2). The similar cross-peak correlations as for 1 observed in the NOESY spectrum gave evidence for the relative stereochemistry of 2 at C-3, C-5, H-9, C-25, and C-26. Combining the comparison of NMR data among 1, 2 and the 18,19-seco-ursane derivatives (Li et al., 2005), compound **2** was established as $(3\beta,12\beta)$ -3-[β -D-glucopyranosyl- $(1 \rightarrow 2)$ - α -L-arabinopyranosyl]-12,19,21-trihydroxy-18,19secours-13(18)-en-28-oic acid, named cornutaoside B. The detailed NMR spectra data of cornutaoside B were listed in Tables 1 and 2.

3.2. Significance of 18,19-seco-ursanes in genus Ilex

The two known compounds were identified as pomolic acid 3-O- β -D-glucopyranosyl $(1 \rightarrow 2)$ - α -L-arabinopyranosyl-28-O- β -D-glucopyranoside (**3**) (Nakanishi et al., 1982) and darutoside (**4**) (Son, Giang, & Taylor, 2005) by spectroscopic analysis, and comparison with literature data. Compound **4** was the first diterpene skeleton isolated from the genus *llex*. To the best of our knowledge, the isolation and characterization of **1** and **2** also showed it to be the first example of an 18,19-*seco*-ursane skeleton from this species, though several 18,19-*seco*-ursane derivatives were previously isolated from *I. aculeolata* (Ouyang, Zhang, & Li, 2002) and *I. crenata* (Kakuno, Yoshikawa, Arihara, 1991; Kakuno, Yoshikawa, Arihara, Takei et al., 1991) with different substitutions at E-ring. The possible biogenetic pathway of 18,19-*seco*-ursanes was recently pro-

Table 2 Selective ¹H NMR data of compounds **1** and **2** (500 MHz, pyridine- d_5)^a

3.33 (1H, dd, <i>J</i> = 11.0, 3.1 Hz)
0.82 (1H, br d, J = 11.6 Hz)
1.48 (2 H, br s)
4.69 (1H, t, J = 6.2 Hz)
6.29 (1H, s)
4.10 (1H, m)
1.93 (1H, m)
5.16 (1H, m)
2.22 (m), 2.41 (m)
1.31 (3H, s)
1.09 (3H, s)
0.90 (3H, s)
1.36 (3H, s)
1.28 (overlap, s)
1.41 (3H, d, <i>J</i> = 6.1 Hz)
1.11 (3H, d, <i>J</i> = 6.5 Hz)
5.05 (1H, s)
4.69 (1H, t, J = 6.2 Hz)
4.45 (m)
4.45 (m)
3.88 (m), 4.41 (m)
5.27 (1H, d, J = 7.7 Hz)
4.19 (1H, t, J = 8.1 Hz)
4.29 (1H, t, J = 9.1 Hz)
4.41 (m)
3.91 (m)
4.52 (2H, br s)

^a Assignments were confirmed by HMQC and HMBC experiments.

posed by Li et al. (2005), in which ursane derivatives of closed E-ring might be converted to 18,19-*seco*-ursanes via a series of oxidation, reduction and rearrangement reactions. In the case of *I. cornuta*, the occurrence of pomolic acid, a predominant aglycone found in leaves, might act as the precursor for this transformation.

3.3. Cytotoxic activities of the 18,19-seco-ursane compounds

Compounds **1** and **2** were evaluated *in vitro* against two mouse cancer cell lines (L1210 and B16) and one human cell line (U937). 5-Fluorouracil (5-Fu) was used as a positive control. Slight inhibition of cell growth, achieving corresponding 23.75%, 19.99%, and 13.89% decreases in L1210, B16 and U937 cell lines at a concentration of 346.25 μ M, was observed with **2**, whereas significant decreases of 73.76%, 25.73%, and 67.87% were observed with 5-Fu. As for **1**, a similar trend was observed for the above three cell lines

with the inhibition values of 23.13%, 11.48% and 12.00% at a concentration of 443.61 μ M. Results from this cytotoxic test showed that compounds **1** and **2** were found to have no potential cytotoxic activities under the current experiment conditions.

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